Title: Synthesis and characterization of cyclometallated complexes of imines: 4-

CIC6H4CH=NCH2(4'-FC6H4) and 2,3-F2C6H3CH=NCH2(4'-FC6H4).

Expanding the scope of cyclopalladations.

Student: Ulunay Ates Ceresuela

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Supervisor: Dr. Granell Sanvicente, Jaume

Departamento de Química Inorgánica i Orgánica

The treatment of 4-fluorobenzylamine with 4-chlorobenzaldehvde 2.3difluorobenzaldehyde in acetone at 80°C afforded the corresponding imines (1), 4-CIC₆H₄CH=NCH₂(4'-FC₆H₄) and 2.3-F₂C₆H₃CH=NCH₂(4'-FC₆H₄) respectively. which were isolated in a yield >90%. The same reaction was performed in milder conditions obtaining similar results. The imines (1) reacted with a stoichiometric amount of Pd(OAc)2 in acetic acid affording the corresponding acetato-bridged five-membered ortho-cyclopalladated dimers (2) in 44-69% yield. The reaction was reproduced in different conditions to improve the yield, obtaining the best results at 85°C for 1h. Compounds 2 were converted by a metathesis reaction with an excess of LiCl and LiBr in acetone into the corresponding chloro- or bromo-bridged dinuclear cyclopalladated compounds. The stoichiometric amount of PPh3 and dppe was added to the halogen-bridge compounds to obtain the mononuclear compounds trans-N,L- $[Pd\{RHC=NCH_2(4'-FC_6H_4)\}(X)(L)]$ [3-A (R= 4-CIC₆H₄, X= Br, L= PPh₃); 3-B (R= 2,3-F₂C₆H₃. X = Br, L = PPh₃); **4-A** (R= 4-ClC₆H₄, X= Cl, L= PPh₃); and the dinuclear compounds **5-A** (R= 4- CIC_6H_4 , X= Br, L= dppe); **5-B** (R= 2,3-F₂C₆H₃, X= Br, L= dppe)] in 50-90% yield. The compounds obtained were characterized by ¹H NMR, ³¹P NMR spectroscopy -in CDCl₃ solution—, elemental analyses, mass spectrometry, and IR spectroscopy.

Some tendencies have been found in this work: i) the *E* isomer of the imine was isolated in all reactions; ii) the *ortho*-fluoro atoms show an interaction with the imine proton; iii) the *endo*-palladacycle is the thermodynamic control isomer; iv) the aromatic protons of the metallated ring are high field shifted in compounds containing phosphines, showing a *trans* arrangement between the phosphines and the nitrogen atom in agreement with the transphobia; v) the methinic proton signal in proton NMR spectra is displaced toward high fields confirming the nitrogen-metal bond.